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# Differential effects of $\text{Cu}^{2+}$ and $\text{Fe}^{3+}$ ions on in vitro amyloid formation of biologically-relevant $\alpha$ -synuclein variants

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**Abstract** Alterations in metal ion homeostasis appear coupled to neurodegenerative disorders but mechanisms are unknown. Amyloid formation of the protein  $\alpha$ -synuclein in brain cells is a hallmark of Parkinson's disease.  $\alpha$ -Synuclein can bind several metal ions in vitro and such interactions may affect the assembly process. Here we used biophysical methods to study the effects of micromolar concentrations of  $\text{Cu}^{2+}$  and  $\text{Fe}^{3+}$  ions on amyloid formation of selected  $\alpha$ -synuclein variants (wild-type and A53T  $\alpha$ -synuclein, in normal and N-terminally acetylated forms). As shown previously,  $\text{Cu}^{2+}$  speeds up aggregation of normal wild-type  $\alpha$ -synuclein, but not the acetylated form. However,  $\text{Cu}^{2+}$  has a minimal effect on (the faster) aggregation of normal A53T  $\alpha$ -synuclein, despite that  $\text{Cu}^{2+}$  binds to this variant. Like  $\text{Cu}^{2+}$ ,  $\text{Fe}^{3+}$  speeds up aggregation of non-acetylated wild-

type  $\alpha$ -synuclein, but with acetylation,  $\text{Fe}^{3+}$  instead slows down aggregation. In contrast, for A53T  $\alpha$ -synuclein, regardless of acetylation,  $\text{Fe}^{3+}$  slows down aggregation with the effect being most dramatic for acetylated A53T  $\alpha$ -synuclein. The results presented here suggest a correlation between metal-ion modulation effect and intrinsic aggregation speed of the various  $\alpha$ -synuclein variants.

**Keywords**  $\alpha$ -Synuclein · Parkinson's disease · Metal ions · Acetylation · Thioflavin T · Amyloid formation

## Introduction

A unifying molecular event in neurodegenerative disorders is the aberrant self-assembly of proteins into amyloid fibers with a hallmark cross- $\beta$ -sheet arrangement. Parkinson's disease (PD) is the second most common neurodegenerative disorder (after Alzheimer's disease) and the most common movement disorder. PD is characterized by widespread deterioration of subcortical structures of the brain, especially dopaminergic neurons in the substantia nigra (Chen et al. 2014). Conformational changes resulting in aggregation of the intrinsically-disordered protein  $\alpha$ -synuclein ( $\alpha$ Syn) into amyloid fibers is directly related to PD (Winner et al. 2011; Galvin et al. 1999). Many

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amyloidogenic proteins (including  $\alpha$ Syn) can bind metals in vitro and systemic as well as tissue imbalances of metal levels are often found to be strongly associated with neurodegenerative disorders including PD (Fink 2006). This implies that metal ions may play pivotal roles in progression of PD and other neurodegenerative disorders (Bush 2000; Bjorklund et al. 2019; Aaseth et al. 2018). Notably, the brain has a high demand for metal ions and, as such, it can be viewed as an organ that concentrates metal ions (Bush 2000; Valiente-Gabioud et al. 2012).

For PD, it has been found that copper levels are decreased in brain tissue but increased in fluids, whereas for iron, the levels are increased in brain cells (Bush 2000). The differences noted vary between brain regions (Bjorklund et al. 2019; Aaseth et al. 2018). Also other metal ion levels are altered in PD and, for example, occupational exposure to high manganese levels have been linked to Parkinsonism, a disorder similar to PD (Bjorklund et al. 2019; Aaseth et al. 2018). Of metal ions interacting with  $\alpha$ Syn, copper ions display the strongest affinity (nanomolar range) (Binolfi et al. 2010). Copper (Cu) in both redox states (oxidized =  $\text{Cu}^{2+}$ ; reduced =  $\text{Cu}^+$ ) can bind to  $\alpha$ Syn (Camponeschi et al. 2013; De Ricco et al. 2015a) and structural features, binding sites, and affinities for the interaction between  $\alpha$ Syn and  $\text{Cu}^{2+}$ , as well as  $\text{Cu}^+$ , have been the focus of many in vitro spectroscopic investigations (Camponeschi et al. 2013; De Ricco et al. 2015a; Binolfi et al. 2006, 2011; Carboni and Lingor 2015). Whereas the high affinity  $\text{Cu}^{2+}$  site involves backbone nitrogen atoms in residues Met1 and Asp2, the dominant  $\text{Cu}^+$  binding site involves the side chain sulfurs of Met1 and Met5 (Moriarty et al. 2014a; Miotto et al. 2015). Also iron in both redox states (oxidized =  $\text{Fe}^{3+}$ ; reduced =  $\text{Fe}^{2+}$ ) can bind to  $\alpha$ Syn but with micromolar affinity [differs between redox states, and between published reports (Peng et al. 2010; Davies et al. 2011a)] and the  $\alpha$ Syn interaction site involves residues 119 to 124 ( $_{119-124}$ DPDNEA $_{124}$ ) in the C-terminal part (Abeyawardhane and Lucas 2019). Many metal ions can accelerate  $\alpha$ Syn amyloid formation, but  $\text{Cu}^{2+}$  has the largest such effect in vitro (Montes et al. 2014; Davies et al. 2016). However, parallel comparisons of metal-ion effects on  $\alpha$ Syn aggregation, at biologically-relevant metal ion concentrations and for biologically-relevant  $\alpha$ Syn variants (such as with N-terminal acetylation), are lacking.

Although biological systems are complex and many factors must be taken into consideration (e.g., redox reactions, protein partners, local concentrations, membranes) to understand PD development in patients (Bush 2000; Bjorklund et al. 2019; Aaseth et al. 2018), it is of fundamental importance to reveal intrinsic features of how metal ions affect  $\alpha$ Syn amyloid formation. Here we compared the effects of  $\text{Cu}^{2+}$  and  $\text{Fe}^{3+}$  on  $\alpha$ Syn amyloid formation, using micromolar concentrations of metal ions and a set of PD-relevant  $\alpha$ Syn variants, in vitro. The obtained results demonstrate a dependency among metal ion identity, concentration and  $\alpha$ Syn variant that appears related to differential  $\alpha$ Syn monomer conformation and thereby intrinsic aggregation speed.

## Materials and methods

### Expression and purification of $\alpha$ Syn variants

Human WT and A53T  $\alpha$ Syn proteins in non-acetylated forms were expressed heterologously in *Escherichia coli* as reported in Werner et al. (2018). In short, plasmids for WT and A53T  $\alpha$ Syn were transformed into BL21 (DE3) (Novagen) cells. The bacteria were first grown to an OD $_{600}$  of 0.6 in Luria broth (LB) containing 100  $\mu\text{g/ml}$  carbenicillin at 37 °C and then induced with 1 mM IPTG (isopropyl  $\beta$ -D-1-thiogalactopyranoside) and grown overnight at 25 °C post induction. The cells were harvested and lysed by sonication on an ice bath in 20 mM Tris–HCl buffer pH 8.0 in the presence of protease inhibitor cocktail (Roche). The lysate after sonication was treated with a universal nuclease (Pierce) for 15 min at room temperature. The lysate was then heat treated at 90 °C in a water bath for 10 min followed by centrifugation at 15,000  $\times g$  for 30 min. The supernatant was then filtered through 0.2  $\mu\text{m}$  filter and loaded on to a pre-equilibrated 5 ml HiTrap Q FF anion exchange column (GE Healthcare). The  $\alpha$ Syn proteins were eluted by a linear gradient with 1 M NaCl in 20 mM Tris–HCl buffer pH 8.0. The eluted protein were run on a 4–12% SDS-PAGE and fractions containing the protein of interest were pooled and concentrated with Amicon Ultra-15 10 K centrifugal filter units (Millipore). The concentrated protein was loaded and retrieved from a pre-equilibrated Hiload 16/600 Superdex 75 pg column (GE Healthcare) with

20 mM Tris–sulfate buffer pH 7.4. For all purified  $\alpha$ Syn variants, the sample purity was confirmed by a single band on SDS-PAGE gel, a single elution peak in size exclusion chromatography, and by mass spectrometry. Fractions containing pure protein were pooled and snap frozen in liquid nitrogen and stored at  $-80\text{ }^{\circ}\text{C}$ . The concentration of WT and A53T  $\alpha$ Syn was determined using  $\epsilon_{280} = 5960\text{ M}^{-1}\text{ cm}^{-1}$ . Acetylated WT and A53T  $\alpha$ Syn proteins were overexpressed by co-transforming the pT7-7  $\alpha$ Syn plasmid with pNatB (a kind gift of D. P. Mulvihill) (Johnson et al. 2013), expressing the yeast *N*-acetyl-transferase NatB gene. After expression, the protein purification procedure was exactly the same as for the non-acetylated  $\alpha$ Syn variants, see above. The co-expression process resulted in the isolation of fully amino-terminal acetylated  $\alpha$ Syn protein, which was confirmed by mass spectrometry.

As described earlier (Horvath et al. 2018), preparation of (non-acetylated) truncated  $\alpha$ Syn (residues 1–97) used a pET3a construct (including a tag with a repressor protein and a His-stretch followed by a Caspase 7 cleavage site) with the gene for truncated  $\alpha$ Syn. The plasmid was transformed into BL21(DE3) (Novagen) cells, which were then grown in similar conditions as for WT  $\alpha$ Syn. The cells were harvested by centrifugation and re-suspended in Buffer A (20 mM Tris–HCl pH 8, 20 mM imidazole and 8 M urea), then sonicated and centrifuged for 30 min at  $15,000\times g$ . The supernatant was filtered through a  $0.2\text{ }\mu\text{m}$  filter and applied to a 5 ml Hi Trap Ni–NTA column (GE healthcare) pre-equilibrated with Buffer A. After washing the column with 5 column volumes of buffer B (20 mM Tris–sulfate pH 7.4, 20 mM imidazole, 50 mM NaCl), truncated  $\alpha$ Syn was eluted with buffer B, with added 250 mM imidazole. The tag was cleaved upon addition of Caspase 7 in the ratio of 1:100 (caspase 7: truncated  $\alpha$ S w/w) in the presence of 5 mM TCEP (tris (2-carboxyethyl) phosphine) and incubated overnight at  $4\text{ }^{\circ}\text{C}$ . The protein was checked for cleavage and dialyzed overnight to remove excess imidazole. The cleaved protein was loaded onto a 5 mL HiTrap Q FF anion exchange column (GE Healthcare) and eluted by a linear gradient of 1 M NaCl in 20 mM Tris–HCl buffer, pH 8.0. The eluted fractions were pooled, concentrated and loaded onto a Hiload 16/600 Superdex 75 column (GE Healthcare). The protein was eluted in 20 mM Tris–HCl buffer pH

7.4. Fractions containing  $\alpha$ Syn were pooled, snap frozen in liquid nitrogen and stored in  $-80\text{ }^{\circ}\text{C}$ .

#### ThT aggregation assay

$\alpha$ Syn amyloid formation reactions were conducted in 96-well half-area transparent bottom plates with a non-binding surface (Corning, CLS3881) with one 2-mm glass bead in each well using a plate reader-incubator instrument (BMG Labtech, Fluostar Optima). The plates were sealed with transparent tape to prevent evaporation and contamination. Measurements performed in TBS (0.05 M Tris–HCl buffer, pH 7.4 with 0.15 M NaCl, 93318 Sigma-Aldrich) buffer in the presence of  $20\text{ }\mu\text{M}$  Thioflavin T (ThT, T3516 Sigma-Aldrich) at  $37\text{ }^{\circ}\text{C}$  using 200 rpm agitation for 5 min during the 20 min measurement cycles, fluorescence was measured from the bottom side of the plate. All  $\alpha$ Syn samples were gel filtered prior to aggregation experiments in order to ensure the starting point is only monomers. In all aggregation measurements, the  $\alpha$ Syn concentration was  $50\text{ }\mu\text{M}$ . Samples were typically incubated for 60 h and fluorescence measured every 20 min. ThT was excited at 440 nm and emission was recorded at 480 nm. All ThT experiments were performed in quadruplicates at each time, and repeated at least three independent times. For seeded ThT aggregation experiments (in which elongation dominate the kinetics) of WT  $\alpha$ Syn as a function of  $\text{Fe}^{3+}$ , preformed  $\alpha$ Syn amyloids were added ( $5\text{ }\mu\text{M}$  monomer concentration) to fresh  $\alpha$ Syn monomers ( $50\text{ }\mu\text{M}$ ) and aggregation experiments, as above, were started.

#### Atomic force microscopy (AFM)

Aggregated samples from ThT experiments were diluted into MilliQ water (10–20 times) and deposited on freshly cleaved mica. After 10 min, the mica was rinsed with filtered Milli-Q water and dried under a gentle nitrogen stream. Images were recorded on an NTEGRA Prima setup (NT-MDT) using a gold-coated single crystal silicon cantilever (NT-MDT, NSG01, spring constant of  $\sim 5.1\text{ N/m}$ ) and a resonance frequency of  $\sim 180\text{ kHz}$ . 256 pixel images were acquired with 0.5 Hz scan rate. Images were analyzed using the WSxM 5.0 software (Horcas et al. 2007). For each sample images were collected in at least three

different  $50 \times 50 \mu\text{m}$  areas, the images shown are representative for each sample.

### Transmission electron microscopy (TEM)

The samples were taken from the incubation mixture from the ThT-aggregation assays after 60–100 h (the samples were considered to be at the end of fibril formation stage), in the presence and absence of metal ions. The samples were placed on glow-charged, formvar coated, carbon-stabilized copper EM grids. Absorption of sample lasted 2 min, followed by staining with 1 percent aqueous phosphotungstate (1% PTA). The grids were then blotted with filter paper and dried before use (Choi et al. 2018). High contrast images were obtained with a TALOS L120C TEM at an accelerating voltage of 120 kV, using a  $4 \times 4$  k CMOS Ceta camera. The magnification was set to  $17,500\times$ – $300,000\times$ .

### Circular dichroism (CD)

CD was measured in the near-UV region (250–700 nm) on Jasco J-810 and Chirascan-CS spectrophotometers using a 1-cm quartz cuvette at 20 °C. Protein samples (ca 100–150  $\mu\text{M}$   $\alpha\text{Syn}$  in the same TBS buffer as in aggregation experiments) were titrated with  $\text{Cu}^{2+}$ , going from 1:0.25 to 1:2 protein:Cu ratios, with approximately 10 min incubation between each addition. Each experiment was repeated three times. The CD data was normalized to mean residue ellipticity.

## Results

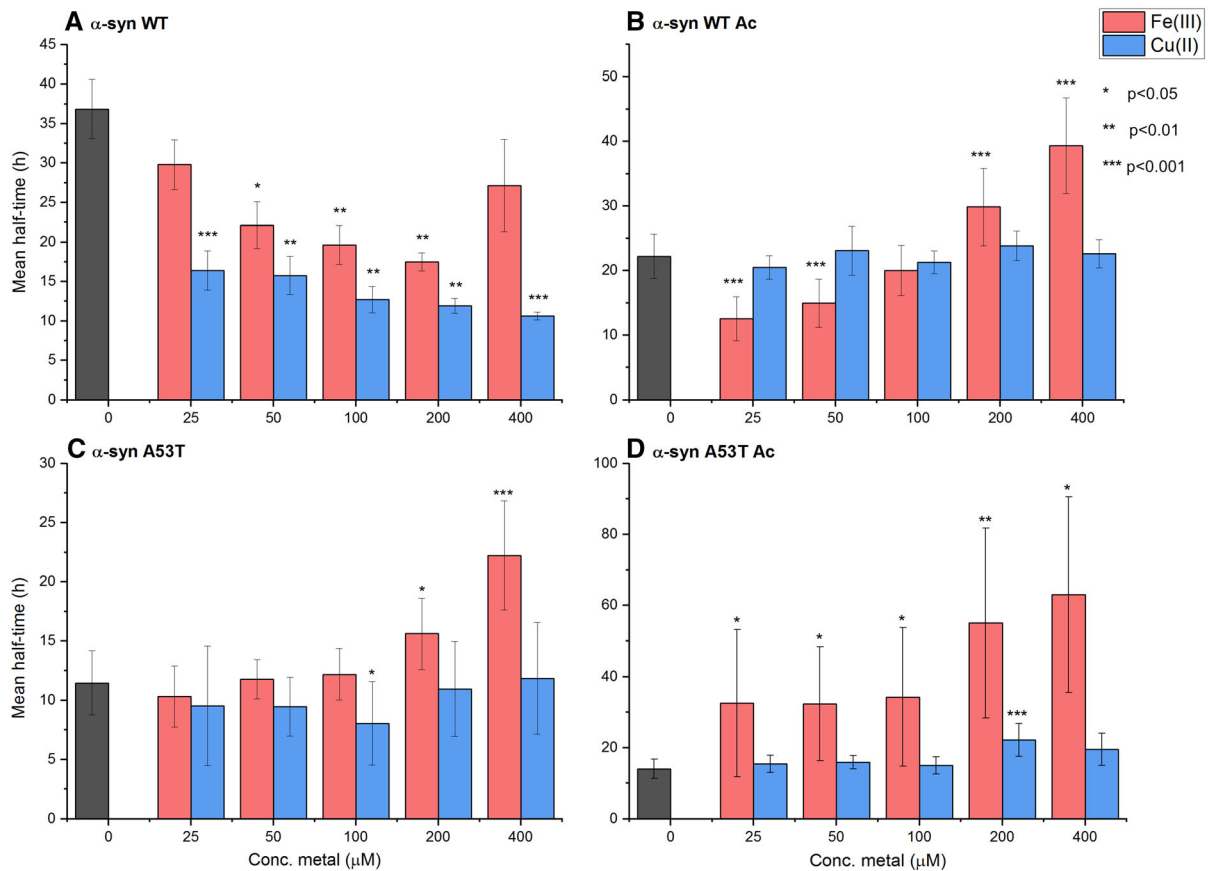
We monitored, using the thioflavin T (ThT) fluorescence assay, the kinetics of aggregation of selected  $\alpha\text{Syn}$  variants (50  $\mu\text{M}$ ) as a function of  $\text{Cu}^{2+}$  and  $\text{Fe}^{3+}$  ions (in the range 0–400  $\mu\text{M}$ ) at pH 7, 37 °C under agitation with a glass bead. Agitation in combination with a glass bead is used to increase the reproducibility of the kinetic data (Giehm et al. 2011). The presence of only  $\alpha\text{Syn}$  monomers at the starting point was ensured by performing gel filtration chromatography of the samples directly prior to each aggregation experiment. As it is now established that  $\alpha\text{Syn}$  is acetylated at the N-terminal amino group in vivo (Bartels et al. 2011), we here focused on wild-type (WT) and Ala53Thr

(A53T; early onset PD-causing mutation) (Polymeropoulos et al. 1997; Kruger et al. 1998)  $\alpha\text{Syn}$  variants in both non-acetylated and acetylated states. The acetylated  $\alpha\text{Syn}$  variants were prepared using an established procedure (Johnson et al. 2013) and modification of the resulting protein was confirmed by mass spectrometry. Representative kinetic aggregation curves for the four proteins alone and upon additions of the metal ions are shown in Figures S1–S3.

### Effect of Cu ions on amyloid formation of $\alpha\text{Syn}$ variants

As expected (Moriarty et al. 2014a; Montes et al. 2014; Davies et al. 2016), we found  $\text{Cu}^{2+}$  to promote aggregation of non-acetylated WT  $\alpha\text{Syn}$  in a concentration dependent manner but these metal ions had no effect on aggregation kinetics of acetylated WT  $\alpha\text{Syn}$  (Figs. 1, S2). Surprisingly, for non-acetylated A53T  $\alpha\text{Syn}$ ,  $\text{Cu}^{2+}$  had no effects on the kinetics of aggregation, although the mutation is not involving Cu-binding residues. In similarity with acetylated WT  $\alpha\text{Syn}$ , and reflecting lack of the high affinity binding site, aggregation of acetylated A53T  $\alpha\text{Syn}$  was not affected by  $\text{Cu}^{2+}$  additions.

To assure that  $\text{Cu}^{2+}$  binds to A53T  $\alpha\text{Syn}$  at our conditions, despite the lack of effect on aggregation kinetics, we turned to near-UV circular dichroism spectroscopy (CD).  $\text{Cu}^{2+}$  binding to the N-terminal  $\text{Cu}^{2+}$  site in  $\alpha\text{Syn}$  can be detected via a negative CD signal around 300 nm (charge transfer transition from metal center to an imidazole group or deprotonated peptide nitrogen) and a positive CD signal at 600 nm (d–d transition) (Binolfi et al. 2006, 2010; Rasia et al. 2005). In Fig. 2, we show that non-acetylated A53T  $\alpha\text{Syn}$  binds  $\text{Cu}^{2+}$ , in an apparent similar coordination to WT  $\alpha\text{Syn}$  (Binolfi et al. 2010), but none of the two acetylated variants bind  $\text{Cu}^{2+}$ . Thus,  $\text{Cu}^{2+}$  binds efficiently to A53T  $\alpha\text{Syn}$  but this interaction does not affect aggregation kinetics. To test if the lack of  $\text{Cu}^{2+}$  effect on aggregation relates to intrinsic speed of aggregation, as A53T  $\alpha\text{Syn}$  aggregates faster than WT  $\alpha\text{Syn}$ , we investigated truncated  $\alpha\text{Syn}$  (contains only residues 1–97) that aggregates even faster than A53T  $\alpha\text{Syn}$ . The CD and ThT data in Fig. S4 demonstrate that truncated  $\alpha\text{Syn}$  interacts with  $\text{Cu}^{2+}$  like WT and A53T  $\alpha\text{Syn}$ , but (in similarity to A53T  $\alpha\text{Syn}$ ) this



**Fig. 1** Midpoints of aggregation curves (derived from ThT-detected kinetic data; Figures S2–S3) for various additions of  $\text{Fe}^{3+}$  (red) and  $\text{Cu}^{2+}$  (blue) to  $\alpha$ Syn variants: **a** WT  $\alpha$ Syn. **b** Acetylated WT  $\alpha$ Syn. **c** A53T  $\alpha$ Syn. **d** Acetylated A53T  $\alpha$ Syn. Significance scale according to: \* $p < 0.05$ , \*\* $p < 0.01$ ,

\*\*\* $p < 0.001$ . The  $p$  values were obtained by the Student's  $t$  test method. The shown data is based on three independent experiments with four replica in each. Midpoints are defined as the time when the ThT signal has reached 50% of its final value

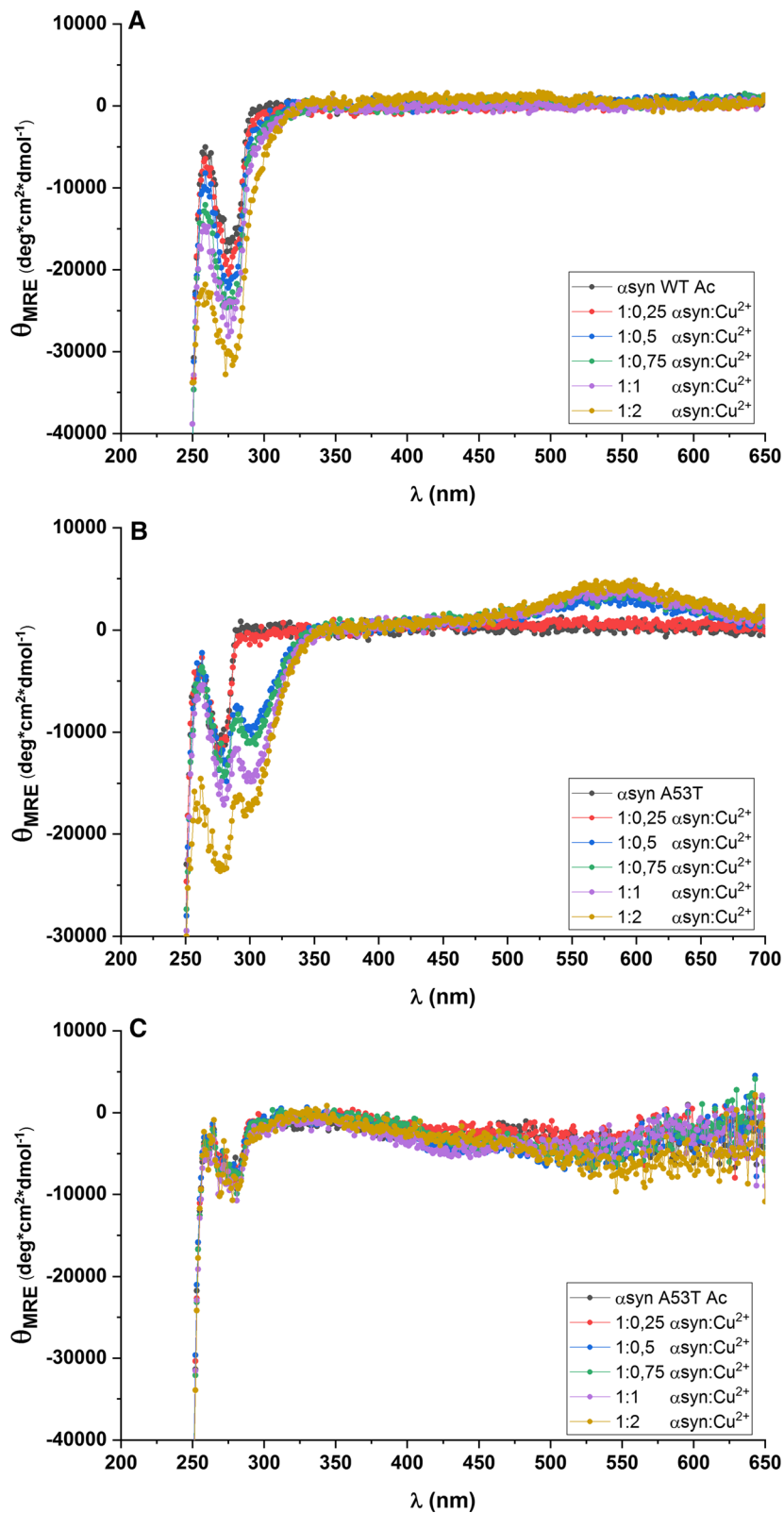
interaction has no effect on aggregation kinetics of truncated  $\alpha$ Syn.

#### Effect of Fe ions on amyloid formation of $\alpha$ Syn variants

Like  $\text{Cu}^{2+}$ ,  $\text{Fe}^{3+}$  speeds up aggregation of non-acetylated WT  $\alpha$ Syn, but with acetylation,  $\text{Fe}^{3+}$  additions gives a complex trend (Figs. 1, S3). At low  $\text{Fe}^{3+}$  concentration (25, 50  $\mu$ M), aggregation of  $\alpha$ Syn becomes faster whereas at higher  $\text{Fe}^{3+}$  concentration (200 and 400  $\mu$ M) metal binding slows down aggregation of  $\alpha$ Syn. In contrast, for A53T  $\alpha$ Syn, regardless of acetylation or not, for all concentrations,  $\text{Fe}^{3+}$  slows down aggregation with the effect being most dramatic for the acetylated form of A53T  $\alpha$ Syn (Figs. 1, S3). We

note that  $\text{Fe}^{3+}$  has a tendency to precipitate in pH-neutral solutions (Peng et al. 2010) and we checked on this carefully. At our conditions and incubation times, like in other studies (Abeyawardhane et al. 2018), there was no precipitation of  $\text{Fe}^{3+}$  detected. In addition, as a control, we tested the effect of  $\text{Fe}^{3+}$  ions on the aggregation reaction of truncated  $\alpha$ Syn. As the binding site for  $\text{Fe}^{3+}$  is suggested to be in the C-terminus (Joppe et al. 2019),  $\text{Fe}^{3+}$  should not affect aggregation of truncated  $\alpha$ Syn—unless artifacts were at play. In Figure S5, we show that  $\text{Fe}^{3+}$  has no effect on truncated  $\alpha$ Syn aggregation kinetics. This result supports that modulation of aggregation kinetics of WT and A53T  $\alpha$ Syn upon  $\text{Fe}^{3+}$  additions are consequences of metal binding to the protein.

**Fig. 2** Near-UV CD spectra for  $\alpha$ Syn variants upon addition of  $\text{Cu}^{2+}$  as indicated. **a** Acetylated WT  $\alpha$ Syn. **b** A53T  $\alpha$ Syn. **c** Acetylated A53T  $\alpha$ Syn





By the use of seeded  $\alpha$ Syn aggregation experiments, where the fiber elongation process dominates, one can assess if an inhibitory effect is due to blockage of amyloid fiber elongation or not (Meisl et al. 2016). Seeding of the aggregation reaction of acetylated WT  $\alpha$ Syn (using 10% amyloid seeds) speeds up aggregation, as expected. When  $\text{Fe}^{3+}$  was added to such seeded reactions, the metal ions had no effect on  $\alpha$ Syn aggregation kinetics (Figure S6). Thus,  $\text{Fe}^{3+}$ -mediated inhibition of amyloid formation of acetylated  $\alpha$ Syn is not due to effects on elongation but appears linked to nucleation steps (primary or secondary nucleation).

For all the  $\alpha$ Syn aggregation reactions in the presence of metal ions, we checked the resulting amyloid fibers by AFM or TEM at the end of the aggregation experiments (Figure S7). In all cases, including the ones where fiber formation was delayed, amyloids are formed and there are no striking differences in amyloid fiber appearances (for this resolution) without and with added metal ions.

## Discussion

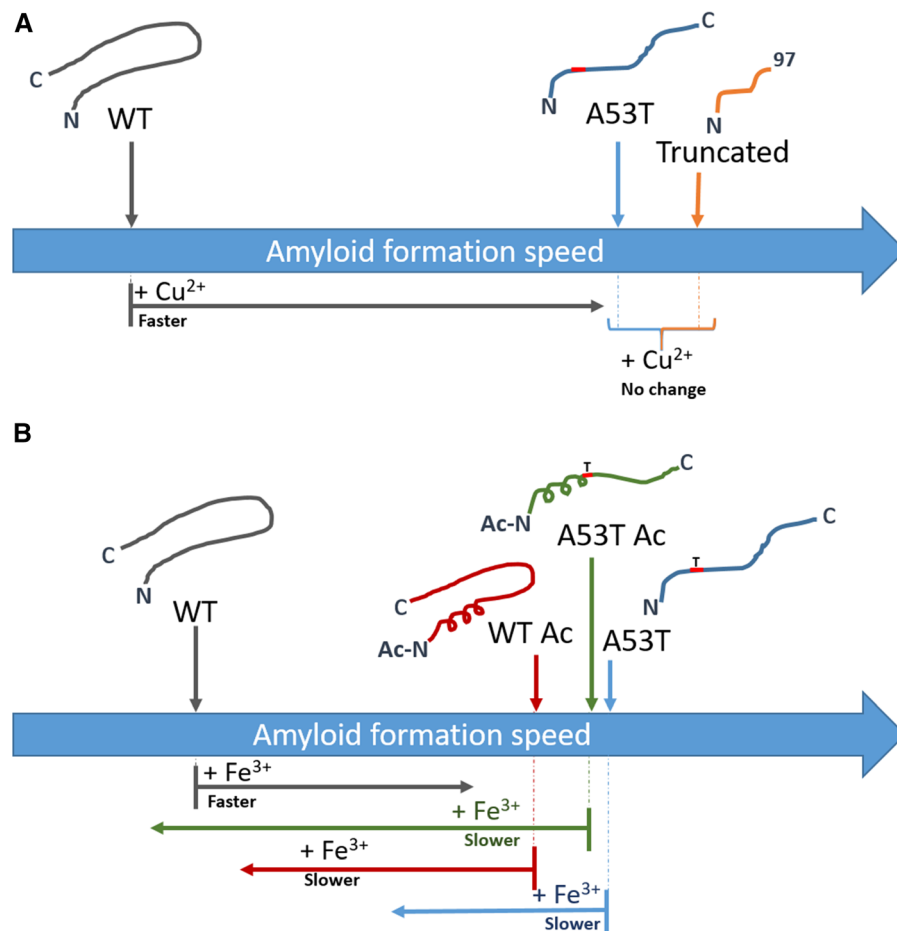
To eventually counteract neurodegenerative diseases, we must gain better knowledge of metal-ion interactions that may modulate amyloid formation. Interestingly,  $\alpha$ Syn, which aggregates into amyloid fibers in PD, is annotated in [uniprot.org](http://uniprot.org) as copper-binding and it can also bind other metal ions in vitro (Montes et al. 2014; Davies et al. 2016). Progressive increase of metal ion content in brain tissue is thought to be a consequence of normal aging and may be one factor promoting age-related neurodegeneration (Valiente-Gabioud et al. 2012). Also prolonged exposure to *e.g.* iron, manganese and copper via external sources (such as industrial work exposure) increases the risk for PD and other neurodegenerative diseases (Bjorklund et al. 2019; Aaseth et al. 2018; Valiente-Gabioud et al. 2012). In brain biopsies from PD patients, iron levels are elevated while cellular copper levels appear reduced (Abeyawardhane and Lucas 2019). Nonetheless, extracellular copper levels at the synapse may reach high micromolar concentrations (Kardos et al. 1989) and for PD patients, copper is elevated in serum (Bush 2000). Because iron and copper dys-homeostasis appears directly related to PD, and both metal ions bind to  $\alpha$ Syn in vitro, we here compared the effects of micromolar concentrations of  $\text{Cu}^{2+}$  and  $\text{Fe}^{3+}$  on the

in vitro aggregation of selected, biologically-relevant  $\alpha$ Syn variants. Our results may act as a biophysical starting point for more advanced investigations taking biological aspects into account. Although there has been previous reports of the effects of these metal ions on  $\alpha$ Syn aggregation, the two ions have not been compared in parallel nor have the consequences of  $\alpha$ Syn mutation and acetylation been addressed in combination with these metal ions.

In accord with several reports, we found  $\text{Cu}^{2+}$  to accelerate non-acetylated WT  $\alpha$ Syn aggregation (Montes et al. 2014; Davies et al. 2016; Uversky et al. 2001) whereas with acetylation, the reaction was not affected by  $\text{Cu}^{2+}$  (Moriarty et al. 2014b; Mason et al. 2016). Surprisingly, for A53T  $\alpha$ Syn, regardless of acetylation or not,  $\text{Cu}^{2+}$  additions did not affect aggregation kinetics although our CD experiments, as expected, showed that  $\text{Cu}^{2+}$  binds to non-acetylated A53T  $\alpha$ Syn like it does to WT  $\alpha$ Syn. For the acetylated forms, WT and A53T  $\alpha$ Syn, no CD signals appeared upon  $\text{Cu}^{2+}$  additions, indicative of lack of metal-ion binding. Acetylation blocks the N-terminal high-affinity  $\text{Cu}^{2+}$  site, but His50 (and C-terminal residues with weak affinity to  $\text{Cu}^{2+}$ ) remains a putative  $\text{Cu}^{2+}$ -binding site. However, if  $\text{Cu}^{2+}$  would bind around His50, such an interaction should give rise to a CD signal (Tian et al. 2019) and thus we conclude that for our acetylated  $\alpha$ Syn variants at our conditions, the affinity of  $\text{Cu}^{2+}$  must be less than 100  $\mu\text{M}$ . That  $\text{Cu}^{2+}$  binding does not affect aggregation kinetics of non-acetylated A53T  $\alpha$ Syn may be related to the intrinsically faster aggregation speed of this variant (Figure S1). The faster aggregation speed of A53T  $\alpha$ Syn (as compared to WT) has been linked to fewer long-range contacts in the mutant's monomeric state (Bertoncini et al. 2005). Thus,  $\text{Cu}^{2+}$  binding to WT  $\alpha$ Syn may change the monomer conformation towards that of A53T  $\alpha$ Syn, with the consequence of faster aggregation (Fig. 3). For A53T  $\alpha$ Syn, already in a more extended conformation,  $\text{Cu}^{2+}$  binding may not affect the conformational landscape of the monomer further, and thus  $\text{Cu}^{2+}$  binding does not affect kinetics of aggregation. In accord with this reasoning, for truncated  $\alpha$ Syn, that aggregates even faster than A53T  $\alpha$ Syn,  $\text{Cu}^{2+}$  binding had no effect on the aggregation kinetics (Figure S4).

It has been reported that millimolar  $\text{Fe}^{3+}$  speeds up non-acetylated WT  $\alpha$ Syn aggregation (Uversky et al. 2001) but the effect of  $\text{Fe}^{3+}$  on aggregation kinetics of





**Fig. 3** Scheme illustrating the link between metal effects and intrinsic  $\alpha$ Syn variant aggregation speed (monomer conformation). **a** Aggregation-promoting effect of  $\text{Cu}^{2+}$  is only found when intrinsic aggregation is slow, as for WT  $\alpha$ Syn. For A53T and truncated (1–97)  $\alpha$ Syn, which both aggregates faster,  $\text{Cu}^{2+}$  has no accelerating effect. **b** Aggregation-modulating effect of  $\text{Fe}^{3+}$  depends on intrinsic aggregation speed. Whereas WT  $\alpha$ Syn is accelerated by  $\text{Fe}^{3+}$ , the faster aggregating variants acetylated

WT  $\alpha$ Syn, A53T  $\alpha$ Syn and acetylated A53T are instead slowed down by  $\text{Fe}^{3+}$ . The intrinsic  $\alpha$ Syn aggregation speed of variants can be linked to their individual monomer conformational landscape, with WT  $\alpha$ Syn being more compacted (involving inhibitory interactions between N- and C-termini) than A53T  $\alpha$ Syn and acetylation results in transient helicity in the N-terminal part (see text)

acetylated  $\alpha$ Syn has not been tested. One study tested the effects of  $\text{Fe}^{3+}$  and  $\text{Fe}^{2+}$  on resulting amyloid fiber structures of acetylated WT  $\alpha$ Syn, but no kinetics were reported (Abeyawardhane et al. 2018). Our data show that when probed in the micromolar range,  $\text{Fe}^{3+}$  accelerates aggregation of non-acetylated WT  $\alpha$ Syn, but for acetylated WT and A53T  $\alpha$ Syn (with and without acetylation) the effect of  $\text{Fe}^{3+}$  is to slow down the aggregation kinetics. We rationalize this by differences in conformations of the monomeric variants that, in turn, relate to intrinsic aggregation speeds. As mentioned, A53T aggregates faster than WT  $\alpha$ Syn

(Figure S1). When WT  $\alpha$ Syn is acetylated, it aggregates faster, but acetylation does not matter much for intrinsic aggregation speed of A53T  $\alpha$ Syn (Figure S1). While the A53T  $\alpha$ Syn variant has less long-range contacts (Bertoncini et al. 2005), N-terminal  $\alpha$ Syn acetylation results in increased local helical content (Kang et al. 2012). Both these alterations thus allows for faster aggregation kinetics, with the mutation having the largest effect. Therefore, we speculate that  $\text{Fe}^{3+}$  binding to non-acetylated WT  $\alpha$ Syn alters the monomer conformation towards that of the mutant (i.e., more extended structure) and this speeds up

aggregation. In contrast, for acetylated WT and both forms of A53T  $\alpha$ Syn,  $\text{Fe}^{3+}$  binding instead changes the conformations (towards more compact forms) such that aggregation becomes slower (illustrated in Fig. 3).

In vivo, many factors, such as metal-induced redox reactions (Abeyawardhane and Lucas 2019; Miotto et al. 2014; Davies et al. 2011b) and metal-binding proteins (Montes et al. 2014; McLeary et al. 2019), have to be taken into account (Bush 2000). For example, it was recently shown that, when bound to  $\alpha$ Syn, copper could cycle between oxidized and reduced forms (Miotto et al. 2014; Davies et al. 2011b; Wang et al. 2010) and such copper-bound  $\alpha$ Syn was able to act as a ferri-reductase using the copper ion as a catalytic redox center (Brown 2013; De Ricco et al. 2015b). In another study, it was shown that the copper transport protein Atox1, which transports  $\text{Cu}^+$  in the cytoplasm, could form a complex with  $\alpha$ Syn (regardless of acetylation) and this interaction blocked  $\alpha$ Syn aggregation (Horvath et al. 2019). It is also possible that the resulting  $\alpha$ Syn amyloid fibers, and intermediate species (oligomers), differ in the presence or absence of metal ions. One study showed that in the presence of  $\text{Cu}^{2+}$ , the formed  $\alpha$ Syn amyloid fibers appeared more neurotoxic than in the absence of  $\text{Cu}^{2+}$  (Choi et al. 2018). To resolve the connection between metal ion metabolism and neurodegeneration, many additional studies, spanning from biophysics in vitro (as here) to cellular models and living systems, are desired.

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### Compliance with ethical standards

**Conflict of interest** The authors declare no competing interests.

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